





Rates of reaction



Topic	We are learning to:		
Chemical reactions	What is the rate of a reaction		
	Explain what happens in a reaction in terms of collisions (Collision Theory)		
Recall how Factors change the rate of a reaction	changing the temperature of a solution At higher temperatures, particles can collide more often and with more energy, which makes the reaction take place more quickly		
	changing the concentration of a solution More reactant particles moving together allow more collisions to happen and so the reaction rate is increased.		
	changing the surface area of a solid more particles are exposed to the other reactant. there is a greater chance of particles colliding, which leads to more successful collisions per second. the rate of reaction increases.		
	adding a catalyst to a reaction increased the frequency of successful collisions as more particles have enough energy to take part in the reaction, therefore there are more successful collisions.		
Experiments	Be familiar with the experiments in this booklet (how rate was measured each time)		
variables	Define control variable, Independent variable, and dependent variable and be able to identify each one in an investigation.		
Graph drawing skills	Label X and Y axis, Draw suitable scale, continuous data -plot points accurately, join points together with a suitable line.		
	Be able to draw a Bar Chart for discrete data		

On the move



Topic	We are learning to:		
Speed Trap	State the equation for Average Speed, distance travelled and time taken.		
	Use the equations (above) to carry out calculations.		
	State the units of Speed.		
Journey Graphs	Interpret a Distance – time graph.		
	Draw a distance – time graph from information given.		
	Use information from a Distance–time Graph to calculate speed.		
	Use a Distance – time graph to describe an objects journey.		
	State the units of Speed.		

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Braking News	State the definition of Thinking Distance.		
	State the definition of Braking Distance.		
	State how to calculate the total braking distance.		
	Write down how total stopping distance can be increased.		
	Interpret a thinking distance/braking distance chart.		



Healthy Body

Topic	We are learning to:		
Food	Know the 7 nutrients necessary for good health		
	Food tests –Protein Sugar Starch Fat		
	Recall the consequences of imbalances in the diet, including obesity, starvation and deficiency diseases Careers – know the work of a dietician		
Digestion	Digestive system – know the main organs and their function		
	chemical digestion - know the definition of an enzyme, the names of 3 digestive enzymes and their role.		
	Mechanical Digestion – Know the structure of teeth Know the different types of teeth and their roles		
Exercise	know the importance of exercise to health		
	the interaction between skeleton and muscles		
	the function of muscles and examples of antagonistic muscles		
	the structure and functions of the human skeleton, to include support, protection, movement and making blood cells		
Drugs	the effects of recreational drugs (including substance misuse) on behaviour, health and life processes.		



Atoms and elements

Lesson	What we learned		
Elements and compound	Sort substances into elements and compounds.		
	Interpret chemical formulae and relate the numbers to the number of atoms involved.		
	Understand energy change during compound formation		
	The formation of iron sulfide from Iron and Sulfur.		
Chemical reactions	Describe how elements can react with the substances around them.		
	Explain that metals often react with the elements in the air to form compounds.		
	Explain that thermal decomposition means to break down a compound using heat energy.		
How do compounds react with each other?	Evidence for a chemical reaction has taking place.		
	Recognise that reactions can take place between elements or compounds.		
	Explain observations in terms of reacting particles.		
mixtures	Classify materials as elements, compounds and mixtures.		
	Explain that mixtures do not have a fixed composition and cannot be represented by a chemical formula		
	Know that particles in a mixture are not chemically joined together.		
What is a pure substance?	Know that elements and compounds melt and boil at a particular temperature.		
	Explain how the melting and boiling points can be used to identify substances.		
	Know that mixtures do not melt or boil at a fixed temperature.		
Solutions	What a solvent is.		
	What a solute is.		
	What a solution is.		
	What dissolving means.		
	What soluble means.		
Separation techniques	How to separate miscible liquids.		
	How to separate pigments.		
	How to separate a soluble substance from an insoluble substance.		
	How to separate pure water from salt water.		
	How to separate Immiscible liquids.		

Gas exchange

Topic	Respiration and Breathing		
respiration	Recall (Aerobic) Respiration word equation glucose + oxygen → carbon dioxide + water + energy		
breathing	The structure and function of gas exchange system in humans (lungs, trachea, bronchi, bronchioles, alveoli, diaphragm, ribs and intercostal muscles)		
	The mechanism of breathing to move air in and out of the lungs		
	Compare inhaled and exhaled air		
	Describe the mechanism of ventilation in terms of changes in volume and pressure		
	Simple measurements of lung volume		
smoking	Describe the consequences of Smoking		
	The impact of exercise, smoking and vaping on the human gas exchange system		
exercise	The effects of exercise on the circulatory system		
	The relationship between heart beat, breathing rate and exercise		

Electricity

Topic	We are learning to:		
Electric Circuits	Describe electric current as the flow of electrons.		
	Describe electric circuits.		
	Recognise a complete working circuit.		
	Recall circuit symbols for common components		
Electric Current	Relate bulb brightness to current.		
Conductors and Insulators	Define the terms conductor and insulator		
	Design and build a circuit to test if a material conducts electricity.		
	Categorize materials as conductors and insulators.		
Resistance	Describe resistance.		
	State the unit and symbol of resistance.		
	Define conductors and insulators in terms of resistance.		
	Give examples of variable resistors in everyday life.		

